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**RESEARCH ARTICLE** 

## Differential Scanning Calorimetry of Polyvinyl Nitrate and Additives

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## ABSTRACT

DSC of fibrous PVN samples, differing in nitrogen content from 11.76% N to 15.71% N, has been carried out at  $5^{\circ}$ C/min heating rate. Results indicate that thermal stability is maximum for PVN containing 11.71% N and decreases with increasing %N. Calorimetric value, however, is maximum (3445 J/g) for fibrous PVN containing 15.72% N. DSC of gelatinized PVN containing 15.71% N shows that its heat of decomposition is 3100 J/g, and this value is generally lowered by incorporating additives, namely, DPA, 2NDPA, carbamite and resorcinol, in 0.25%-1% weight concentration. Also, these additives do not significantly affect the onset temperature and the peak temperature.

*Keywords:* Fibrous, Thermal Stability, calorimetric value, Gelatinized, Heat of decomposition, Additives, on set temperature, Peak temperature

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## INTRODUCTION

The results on TGA and thermal decomposition kinetics of fibrous as well as gelatinized samples of polyvinyl nitrate (PVN) have been published [1]. To augment these studies, we are reporting, in this paper, the results of our work on differential scanning calorimetry (DSC) of the following samples of PVN:-

- a) Fibrous PVN, containing 11.76% N, 13.34% N, 14.95% N and 15.71% N.
- b) Gelatinized PVN, containing 15.71% N, without and with additives.

## MATERIAL AND METHODS

#### Materials

Fibrous PVN, with mol. wt. about 100,000 and nitrogen content between 11.76% and 15.71%, was prepared by controlled nitration of polyvinyl alcohol [2]. PVN (15.71%N) was gelatinized with acetone, and during gelatinization, four different additives, namely diphenylamine (DPA), 2-nitrodiphenylamine (2NDPA), carbamite and resorcinol, were separately incorporated so as to obtain additive concentration of 0.25%, 0.5%, 0.75% and 1% by weight [2].

## Method

DSC of the test samples was carried out using Dupont Instruments 910 DSC apparatus. 1-3 mg samples were used for the experiments. A heating rate of  $5^{\circ}$ C/min and a constant flow rate of nitrogen gas (20ml/min) were maintained.

## **RESULTS AND DISCUSSION**

## DSC of Fibrous PVN (11.76%N-15.71%N)

DSC thermograms for the four samples of fibrous PVN, differing in %N, are shown together (to facilitate comparative assessment) in Diagram 1. The data on exothermic onset temperature, peak temperature and calorimetric value for the same PVN samples are presented in Table 1.

Rate o C/ min. Sample Size 1 of o mg/							
Sample	Exothermic onset temperature 0ºC	Peak temperature 0ºC	Calorimetric Value J/g				
PVN (11.76%N)	190.5	205.0	1533				
PVN (13.34%N)	182.3	195.9	1991				
PVN (14.95%N)	185.2	196.9	3070				
PVN (15.71%N)	181.2	196.3	3445				

**Table-1:** DSC Results of Fibrous PVN samples Containing Different %N. (Heating<br/>Rate=5°C/min. Sample size 1 of 3 mg)

DSC results confirm that the thermal properties of fibrous PVN are largely dependent upon %N in the samples. For the samples having 13.34% N, 14.95% N and 15.1%N, the exothermic onset temperature and the peak temperature are  $183.3\pm2^{\circ}$ C and  $196.4\pm0.5^{\circ}$ C, respectively, but these temperatures significantly increase to  $190.5^{\circ}$ C and  $205^{\circ}$ C, respectively, for the sample having 11.76%N. Hence, fibrous PVN with 11.76% N is thermally morestable than the others, probably because it has relatively less number of the weaker O-NO2, bond (bond energy 57 kcal/mol) in the polymer chain and is, therefore, less favourable to auto-analysed decomposition.

Calorimetric value of fibrous PVN increases considerably, that is, from 1533 J/g to 3445 J/g, with %N increasing from 11.76% to 15.71%. This happens because, with increasing degree of nitration, the oxygen balance also increases leading to higher reaction rate and grater exothermicity of decomposition.

### DSC of gelatinized PVN (15.71%N) without and with Additives

DSC thermograms for gelatinized neat PVN (15.71%N) is shown diagram 2, whereas two DSC thermograms for gelatinized PVN (15.71%N), doped with 0.25% carbamite and with resorcinol, are shown together in diagram 3. The data on exothermic onset temperature, peak temperature and calorimetric value for gelatinized PVN (15.71%N), without and with additives, are presented in Table 2.

Sample Composition			Exothermic	Deel-	O a la mirma atmi a
PVN type	Additive	Conc. of additive %	onset temperature 0ºC	temperature 0°C	Value J/g
PVN (15.71%N), * Fibrous PVN (15.71%N), gelatinized	NONE	0.00	181.6	196.3	3445
	NONE	0.00	185.8	196.1	3100
	DPA	0.25	185.4	196.7	2746
		0.50	188.7	196.0	2783
		0.75	184.2	196.8	2818
		1.00	171.9	194.7	3034
	2NDPA	0.25	185.9	198.2	2700
		0.50	185.6	197.9	3113
		0.75	183.4	196.3	2835
		1.00	185.0	197.0	2798
	Carbamite	0.25	183.0	196.3	2873
		0.50	185.2	197.5	2861
		0.75	185.3	197.4	2773
		1.00	184.0	196.5	3052
	Resorcinol	0.25	185.1	196.8	2966
		0.50	184.8	197.1	2768
		0.75	185.9	194.5	2696
		1.00	186.8	106.3	3036

Table-2: DSC Results on	Gelatinized PVN (15.71%N) without and with Additives. (Heati	ing
	Rate=5°C/min. Sample size 1 of 3 mg)	

\*Included for comparison only.











It is observed from these data that the calorimetric value of fibrous PVN (15.71%N) decreases on gelatinization, from 3445 J/g to 3100 J/g. The calorimetric vale of gelatinized PVN (15.71%N) is usually further lowered by incorporation of the additives, namely, DPA, 2NDPA, carbamite and resorcinol, in 0.2% to 1% concentration, with very few exception. This may be due to the relatively nonenergetic nature of the additives. It is also observed that these additives do not significantly affect the onset temperature and the peak temperature values.

#### CONCLUSION

Thermal stability of fibrous PVN decreases with increasing %N. However, the heat of decomposition of fibrous PVN increases from 1533 J/g to 3445 J/g, corresponding to increase in %N from 11.76 to 15.71. Heat of decomposition of PVN (15.71% N) decreases, on gelatinization, to 3100 J/g, and usually decreases further on incorporating the additives DPA, 2NDPA, carbamite and resorcinol in 0.25%-1% concentration.

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