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## **ORIGINAL ARTICLE**

# *Ex-situ* Management of Ammonium Contamination in Water using Eight Aquatic Plants

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## ABSTRACT

The study was conducted to determine the ammonium  $(NH_{4^+})$  level in drinking and surface waters of Lucknow and to assess the  $NH_{4^+}$  accumulation potential of 8 aquatic plants. Ammonium level in drinking water was higher than permissible limit (PL) i.e. 0.5 mg L<sup>-1</sup>. A few urban sites showed 6-8 folds high  $NH_{4^+}$  over the PL. Ammonium in the periurban and rural site was comparatively lower, though it was still 2-4 folds higher than PL. On the other hand, all surface water sites, except one residential site (Aashiana), showed  $NH_{4^+}$  level less than PL (5 mg L<sup>-1</sup>). Eight aquatic plants viz. Peltandra virginica, Utricularia vulgaris, Eichhornia crassipes, Trapa natans, Mimulus glabratus, Marsilea quadrifolia, Pistia stratiotes and Polygonum persicaria were investigated and cultured in 10 L plastic pots for 15 days (d) under simulated net house conditions. Reduction in  $NH_{4^+}$  level in water and its simultaneous increase in plant tissues was recorded at 5<sup>th</sup>, 10<sup>th</sup> and 15<sup>th</sup> d. Pistia stratiotes was found more efficient and removed  $NH_{4^+}$  upto 70%. Highest accumulation of  $NH_{4^+}$  i.e. 46% was noticed in the tissues of Pistia stratiotes.

Key words: Aquatic plants, Drinking water, Ammonium, Phytoremediation, Pistia stratiotes, Surface water.

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## INTRODUCTION

Ammonium (NH<sub>4</sub><sup>+</sup>) is an important nitrogenous pollutant in soil and aquatic systems, and has been found frequently in the water sources [1]. It is the major end product of protein catabolism and major excretory product which remains in the form of unionized (NH<sub>3</sub>) and ionized (NH<sub>4</sub><sup>+</sup>) ammonia. Ammonia is a critical water quality parameter and toxic to aquatic life, but the NH<sub>4</sub><sup>+</sup> is harmless except in extremely high concentrations. At higher concentration NH<sub>4</sub><sup>+</sup> can damage aquatic life, by declining the oxygen level below 2 mg L<sup>-1</sup>, as it oxidizes to nitrite (NO<sub>2</sub><sup>-</sup>) and nitrate (NO<sub>3</sub><sup>-</sup>) [2], and can induce a number of physiological disorders, when its level in the organism is high [3]. Most of NH<sub>4</sub><sup>+</sup> in surface waters is oxidized to NO<sub>2</sub><sup>-</sup> by *Nitrosomonas* bacteria. The presence of NO<sub>2</sub><sup>-</sup> in the environment is a potential threat due to its well documented toxicity to animals [4]. The major metabolites of NO<sub>2</sub><sup>-</sup> are nitric oxide and nitrosamine. Latter one is highly carcinogenic and associated with a high risk of stomach, liver and esophagus cancer [5].

Ammonium removal is accomplished through biological nitrification in which it is converted into NO<sub>2</sub><sup>-</sup> and NO<sub>3</sub><sup>-</sup> by nitrifying bacteria called nitrifers. On the other hand, anammox (Anaerobic Ammonium Oxidation) bacteria co-function with autotrophic nitroso-bacteria *via* the following route NH<sub>4</sub><sup>+</sup> + 1.32 NO<sub>2</sub><sup>-</sup> +  $0.13H^+ \rightarrow 1.02N_2 + 0.26NO_3^- + 2.03H_2O$ . This pathway of NH<sub>4</sub><sup>+</sup> removal is known as 'Completely Autotrophic Nitrogen removal Over Nitrite' (CANON) pathway [6].

Phytoremediation is a rapidly expanding area of environment science that holds great promise for cleaning up the polluted and contaminated environment [7]. There have been a number of reports using the aquatic plants in phytoremediation technology [8]. Because of its low cost and lack of technical problems, phytoremediation has globally become very popular. This technique is currently being studied for potential use in waste water treatment [9]. The high productivity and nutrient removal capability of aquatic plants have created substantial interest for waste water treatment [10, 11].

The present study was intended to find out the extent of  $NH_{4^+}$  pollution in ground and surface waters of Lucknow during various seasons in the repeated year, 2007-2008 and an unconventional access was taken to evaluate the ability of eight aquatic plants.

## **MATERIALS AND METHODS**

**Water Sample:** The drinking water samples of hand pumps were collected from 22 sites of urban, periurban and rural vicinities of Lucknow during 2007-2008. Simultaneously, surface water samples were also collected from 11 different surface water bodies i.e. ponds, lake and river, randomly at uniform depth of 5 cm. Six samples from each site were collected and analyzed for  $NH_{4^+}$  concentration on the same day to avoid the effect of storage and preservatives.

**Plant Culture:** Aquatic plants viz. *Peltandra virginica, Utricularia vulgaris, Eichhornia crassipes, Trapa natans, Mimulus glabratus, Marsilea quadrifolia, Pistia stratiotes* and *Polygonum persicaria* were procured from the different surface water bodies, during the month May and June, 2009 (min./max. temp-24.6/42.4°C). Plants having biomass i.e. 100 g were placed in 3 L plastic pots (4 replicates per treatment), in the net house for 15 days (d). Initial NH<sub>4</sub><sup>+</sup> concentration i.e. 10 mg L<sup>-1</sup> in water was maintained by employing requisite amount of ammonium chloride (NH<sub>4</sub>Cl) salt. At every 5<sup>th</sup> d water/plant samples were collected from the treatment pots. Prior to sampling deionized water was added to the pots to replenish the water lost by evaporation and to regain the initial level.

**Sample Analysis:** Ammonium in the water and plants samples was assayed by the method as described by Herbert *et al.*[12], using Nessler's reagent ( $K_2HgI_4$ ) with the help of UV-1601, UV-visible spectrophotometer (Shimadzu) and Varian, carry 100 Bio, UV-visible spectrophotometer, by preparing standard curve.

**Statistical Analysis:** The data was analyzed using one way ANOVA (Analysis of Variance). The difference between treatments was considered significant at  $p \le 0.05$ .

## **RESULTS AND DISCUSSION**

**Drinking Water:** In the summer 2007-2008, the level of  $NH_{4^+}$  in drinking water was found in the range of 0.18 to 4.08 mg L<sup>-1</sup>. In rainy season it ranged from 0.1 to 3.75 mg L<sup>-1</sup>, while during winter it varied from 1.12 to 4.51 mg L<sup>-1</sup>. The highest value of  $NH_{4^+}$  (4.51 mg L<sup>-1</sup>) that was beyond 8 folds over the Permissible Limit (PL) i.e. 0.5 mg L<sup>-1</sup> [13] was estimated in an urban site (Hazratganj, a business centre of the city) during winter-2008 and lowest (0.10 mg L<sup>-1</sup>) during rainy season-2007 in a peri-urban site (Saleh nagar, a developing area). Later on during 2008, latter one (Saleh nagar) showed more than 2 fold higher  $NH_{4^+}$  level than the PL, which was in accordance with the urbanization resulting in the increase of sewage water content (table 1). Ammonium in peri-urban and rural regions was comparatively lower than that in urban region, though it was still 2-4 folds higher than the PL at most of the sites (table 1). The annual variation was not very significant and seasonal variation was found above the PL however, its level was often found higher in winter season as compared to summer and rainy season that may be due to the continuous leaching and percolation after rainy season to the ground water table [14].

**Surface Water:** In surface water samples, low NH<sub>4</sub><sup>+</sup> level was detected, that was below the PL i.e. 5 mg L<sup>-1</sup> [15]. At only one urban site (Aashiana, a residential area), NH<sub>4</sub><sup>+</sup> level was estimated between 5.40 to 6.65 mg L<sup>-1</sup> during the repeated year of 2007-2008, that was beyond but close to the PL. Ammonium level during summer, rainy and winter season (2007-2008) varied from 1.10 to 5.66, 1.71 to 6.52 and 1.43 to 6.02 mg L<sup>-1</sup> respectively (table 2). Data revealed that maximum elevation in NH<sub>4</sub><sup>+</sup> level was during rainy season in surface water bodies that may be due to the receiving of surplus quantity of agricultural, human and animal waste during the period [11, 16]. In most cases, the level of NH<sub>4</sub><sup>+</sup> was higher in 2008 as compared to that in 2007, which indicated yearly increase in NH<sub>4</sub><sup>+</sup> level in the water (table 2). The values were almost consistent in two different years and in the three different seasons, and the minor variations observed were significant only in few cases.

mg NH <sub>4</sub> + L·1 water							
Sites	Summer		Rainy season		Winter		
	2007	2008	2007	2008	2007	2008	
Urban regions							
Hazratganj	$3.71 \pm 0.40$	$4.08 \pm 0.28$	$3.54 \pm 0.24$	3.75 ± 0.27	3.93 ± 0.27	4.51 ± 0.29	
Kesharbagh	$2.47 \pm 0.20$	$3.15 \pm 0.22$	$2.28 \pm 0.15$	$2.64 \pm 0.20$	$2.98 \pm 0.19$	4.09 ± 0.27#	
Charbagh	$2.60 \pm 0.11$	$2.80 \pm 0.15$	$2.34 \pm 0.13$	$2.90 \pm 0.24$	$2.82 \pm 0.23$	$3.33 \pm 0.23$	
Zoo	$2.45 \pm 0.14$	$2.75 \pm 0.19$	$2.56 \pm 0.20$	$2.59 \pm 0.25$	$2.58 \pm 0.26$	$2.98 \pm 0.28$	
Mall avenue	$2.50 \pm 0.25$	$3.25 \pm 0.23$	$1.97 \pm 0.24$	$2.78 \pm 0.28$	$3.45 \pm 0.28^{*}$	4.22 ± 0.31#	
Ruchi khand	$2.30 \pm 0.50$	$2.98 \pm 0.38$	$1.65 \pm 0.25$	2.51 ± 0.27	$2.23 \pm 0.27$	$3.17 \pm 0.30$	

**Table 1** Seasonal variation of ammonium (NH4+) level in drinking water samples of hand pumps in urban,<br/>peri-urban and rural regions of Lucknow, during 2007-08

Rajni khand	$1.50 \pm 0.20$	$2.36 \pm 0.15$	$1.13 \pm 0.08$	$1.78 \pm 0.14$	$2.24 \pm 0.12^{*}$	$2.64 \pm 0.18$
Gomti nagar	$1.30 \pm 0.20$	$2.05 \pm 0.12$	$1.08 \pm 0.07$	$1.53 \pm 0.13$	$1.87 \pm 0.12$	$2.14 \pm 0.15$
Nishatganj	$1.40 \pm 0.17$	$1.98 \pm 0.13$	$1.17 \pm 0.07$	$1.71 \pm 0.12$	$2.09 \pm 0.11^{*}$	$2.08 \pm 0.15$
Triveni nagar	$1.30 \pm 0.16$	$2.10 \pm 0.11$	$1.16 \pm 0.07$	$1.58 \pm 0.10$	$1.93 \pm 0.09$	$2.07 \pm 0.13$
Aliganj	$1.40 \pm 0.26$	$2.22 \pm 0.14$	$1.23 \pm 0.08$	$1.67 \pm 0.13$	$1.72 \pm 0.10$	$2.34 \pm 0.16$
Peri-urban regio	ons					
5						
Telibagh	$1.60 \pm 0.20$	$1.93 \pm 0.15$	$1.49 \pm 0.12$	$1.89 \pm 0.14$	$1.98 \pm 0.14$	$2.43 \pm 0.17$
Bhadrukh	$1.50 \pm 0.10$	$2.13 \pm 0.21$	$1.54 \pm 0.18$	$1.98 \pm 0.19$	$2.05 \pm 0.22$	3.04 ± 0.24#
Qila	$1.30 \pm 0.06$	$2.24 \pm 0.14$	$1.16 \pm 0.09$	$1.35 \pm 0.14$	$2.18 \pm 0.13^{*}$	$2.57 \pm 0.17$
PGI	$1.40 \pm 0.20$	$1.88 \pm 0.10$	$1.26 \pm 0.05$	$1.77 \pm 0.11$	$2.01 \pm 0.09$	$2.31 \pm 0.13$
Saleh nagar	$0.18 \pm 0.07$	$1.06 \pm 0.13$	$0.10 \pm 0.09$	$0.87 \pm 0.08$	$1.12 \pm 0.12^{*}$	$1.36 \pm 0.11$
BBAUniversity	$1.30 \pm 0.14$	$1.48 \pm 0.10$	$1.16 \pm 0.07$	$1.29 \pm 0.11$	$1.56 \pm 0.11$	$1.67 \pm 0.11$
Rural regions						
Beli kala	$1.80 \pm 0.30$	$1.98 \pm 0.12$	$1.71 \pm 0.11$	$1.73 \pm 0.10$	$2.12 \pm 0.13$	$2.23 \pm 0.15$
Bijnour	$1.30 \pm 0.07$	$2.18 \pm 0.11$	$1.33 \pm 0.07$	$2.22 \pm 0.10$	$2.33 \pm 0.10^{*}$	$2.53 \pm 0.13$
Chiraiyabagh	$1.30 \pm 0.08$	$1.95 \pm 0.10$	$1.21 \pm 0.08$	$1.82 \pm 0.11$	$1.87 \pm 0.09$	$1.99 \pm 0.13$
Sabha khera	$1.20 \pm 0.25$	$1.67 \pm 0.13$	$1.31 \pm 0.07$	$1.53 \pm 0.13$	$1.75 \pm 0.12$	$2.13 \pm 0.15$
Aurangabad	$1.60 \pm 0.17$	$1.72 \pm 0.12$	$1.36 \pm 0.08$	$1.75 \pm 0.11$	$1.55 \pm 0.11$	$1.98 \pm 0.14$
Values are mon	n of 6 replicat	$-\infty + SD$ signi	ficance at $n < 0.0$	5		

Values are mean of 6 replicates  $\pm$  S.D., significance at p  $\leq$  0.05.

**Table 2** Seasonal variation of ammonium (NH4+) level in surface water samples of ponds, lake and river in<br/>urban, peri-urban and rural regions of Lucknow, during 2007-08

	mg NH <sub>4</sub> + L·1 water						
	Summer		Rainy season		Winter		
Sites	2007	2008	2007	2008	2007	2008	
Urban regions							
Gomti river	$2.60 \pm 0.10$	2.99 ± 0.16	$3.48 \pm 0.18$	$3.91 \pm 0.19$	$2.91 \pm 0.15$	$3.22 \pm 0.18$	
Aashiana pond	$5.40 \pm 0.30$	$5.66 \pm 0.22$	$6.65 \pm 0.26^*$	$6.52 \pm 0.27$	$5.55 \pm 0.23$	$6.02 \pm 0.24$	
Aliganj pond	$2.20 \pm 1.0$	$2.76 \pm 0.13$	$3.18 \pm 0.14$	3.85 ± 0.18#	$2.71 \pm 0.13$	$3.56 \pm 0.15$	
Fazullaganj pond	$2.01 \pm 0.30$	$2.31 \pm 0.15$	$2.91 \pm 0.17$	$3.26 \pm 0.20$	$2.36 \pm 0.16$	$3.12 \pm 0.16$	
Bharat nagar pond	$2.20 \pm 0.40$	$3.01 \pm 0.16$	$4.09 \pm 0.18^{*}$	$3.97 \pm 0.19$	$2.91 \pm 0.17$	$3.45 \pm 0.17$	
Peri-urban regions							
Qila pond	$2.60 \pm 0.70$	$2.81 \pm 0.18$	$3.15 \pm 0.20$	3.80 ± 0.22#	$2.86 \pm 0.18$	$3.65 \pm 0.19$	
B.B.A.University lake	$1.80 \pm 0.50$	$2.26 \pm 0.14$	$2.50 \pm 0.16$	$2.83 \pm 0.17$	$2.17 \pm 0.13$	$3.44 \pm 0.14$	
Telibagh pond	$2.0 \pm 0.20$	$2.27 \pm 0.16$	$2.45 \pm 0.18$	$2.91 \pm 0.19$	$2.24 \pm 0.16$	$2.42 \pm 0.17$	
Sharda naher	$1.10 \pm 0.08$	$1.39 \pm 0.25$	1.71± 0.26	$1.95 \pm 0.27$	$1.43 \pm 0.24$	$1.57 \pm 0.24$	
Rural regions							
Devi khera pond	$2.60 \pm 0.70$	$3.11 \pm 0.23$	$3.35 \pm 0.24$	$3.77 \pm 0.26$	$3.08 \pm 0.22$	$3.43 \pm 0.24$	
Aurangabad pond	$2.50 \pm 0.80$	$2.95 \pm 0.19$	$3.18 \pm 0.20$	$3.34 \pm 0.23$	$2.97 \pm 0.19$	$3.17 \pm 0.21$	

Values are mean of 6 replicates  $\pm$  S.D., significance at p  $\leq$  0.05. \*significant values of 2007, #significant values of 2008

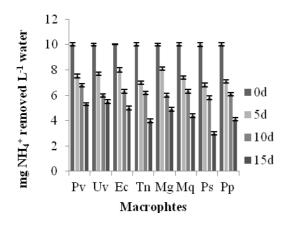


Fig. 1(a) Ammonium removed from water at  $5^{th}$ ,  $10^{th}$  and  $15^{th}$  d.

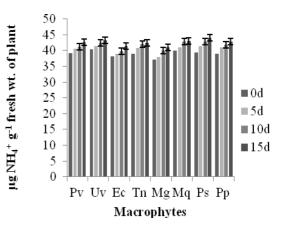


Fig. 1(b) Ammonium accumulated in the plants at  $5^{\text{th}}$ ,  $10^{\text{th}}$  and  $15^{\text{th}}$  d

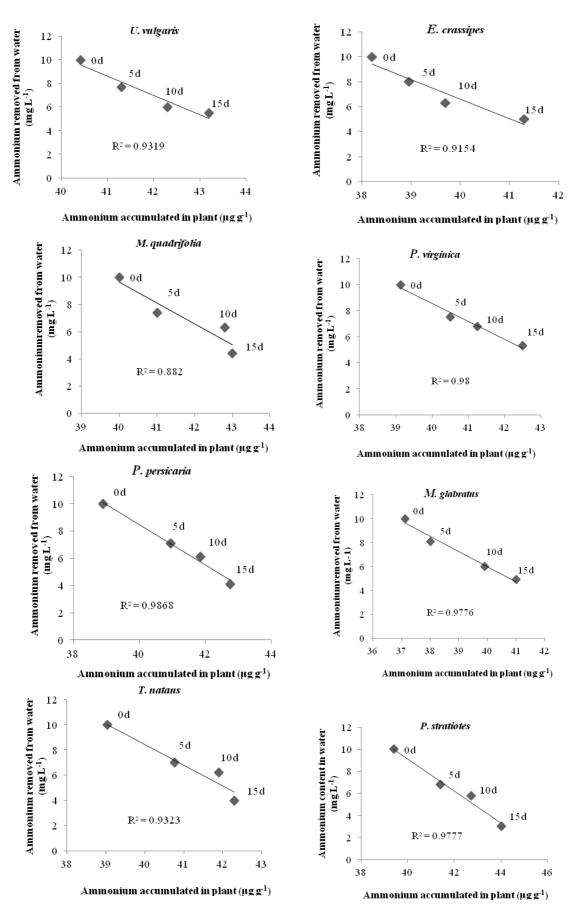


Fig. 2 Linear correlation between ammonium removed from water and accumulated in the plants at 5th, 10th and 15thd.

**Ammonium Remediation:** Ammonium removal from water and accumulation in the plants water was measured by 8 native aquatic plant species. Initial concentration of NH<sub>4</sub><sup>+</sup> employed to the pots containing aquatic plants was 10 mg L<sup>-1</sup>, and at the end of experimental period its level was found in between 3 to 5.5 mg L<sup>-1</sup> in the treated water (fig. 1a). The percent of NH<sub>4</sub><sup>+</sup> removed from the water at 15<sup>th</sup> d by the selected aquatic plants ranged from 45%-70% (fig. 1a). Content of NH<sub>4</sub><sup>+</sup> in the plants tissues before plantation was in between 37.12-40.42 µg g<sup>-1</sup>. At 15<sup>th</sup> d, NH<sub>4</sub><sup>+</sup> estimated in the different plants was found in between 41-44.01 µg g<sup>-1</sup> (fig. 1b).

The highest content of  $NH_{4^+}$  i.e. 4.6 µg g<sup>-1</sup> equivalents to 46% was accumulated in the tissues of *Pistia stratiotes* whereas least i.e. 3.88 µg g<sup>-1</sup> or 38.8%  $NH_{4^+}$  was accumulated in the tissues of *Mimulus glabratus* at 15<sup>th</sup> d (fig. 1b). On the basis of above findings it is observed that the amount of  $NH_{4^+}$  reduced from the water was 6.2-24% more than that accumulated in the different aquatic plants. This difference may be due to the oxidation of ammonium into  $NO_2^-/NO_3^-$  by nitrifying/denitrifying aerobic/anaerobic microorganisms [6]. Results indicated that  $NH_{4^+}$  content increased linearly at 5<sup>th</sup>, 10<sup>th</sup>, and 15<sup>th</sup> d in the plants tissues with simultaneous decrease in its level in water (fig. 2).

All the selected aquatic plants accumulated NH<sub>4</sub><sup>+</sup> from the water. However, NH<sub>4</sub><sup>+</sup> content was found greater in *Pistia stratiotes* which is similar to other studies [17]. *Pistia stratiotes* is a cosmopolitan species and grows round the year in nutrient enriched water. This aquatic plant may also be recommended for the NH<sub>4</sub><sup>+</sup> phytoremediation from surface water because of lower gain of mass, thus reducing problems from exceeded biomass [17, 18].

## CONCLUSIONS

The result of this study indicated that ground water  $NH_{4^+}$  level was more than 8 folds over the PL that may pose threat to humans. On the other hand all the surface water bodies exhibited  $NH_{4^+}$  level which was below the PL, except one site Aashiana of urban region. By the analysis of data presented in this paper, we can conclude that *Pistia stratiotes* can be utilized as a hyper-accumulator of  $NH_{4^+}$  and can phytoremediate it upto 70%.

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